

Faculty of Manufacturing Engineering

PREPARATION OF GRAPHENE/MOLYBDENUM DISULFIDE BASED ELECTRODES AND ITS ELECTROCHEMICAL PERFORMANCE IN SUPERCAPACITORS

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A thesis submitted in fulfilment of the requirements for the degree of Doctor of Philosophy

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Faculty of Manufacturing Engineering

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UNIVERSITI TEKNIKAL MALAYSIA MELAKA

2019

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DECLARATION

I declare that this thesis entitled "Preparation of Graphene/Molybdenum Disulfide Based Electrodes and Its Electrochemical Performance in Supercapacitors" is the result of my own research except as cited in the references. The thesis has not been accepted for any degree and is not concurrently submitted in candidature of any other degree.

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APPROVAL

I hereby declare that I have read this thesis and in my opinion this thesis is sufficient in terms of scope and quality for the award of Doctor of Philosophy.

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DEDICATION

To my beloved family.

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ABSTRACT

Supercapacitor is highly promising energy device due to its electrical charge storage performance and significant lifecycle ability. Construction of the supercapacitor cell especially its electrode fabrication is critical to ensure great application performance. The purpose of this research is to fabricate the molybdenum disulfide (MoS₂), graphene and G/MoS₂ hybrid electrode and their usage as symmetric and asymmetric supercapacitors. The electrode was prepared by using a simple and facile slurry technique. By this, XRD was used to analyze the crystal phase and structure of the as-prepared graphene, MoS₂, and G/ MoS₂ hybrid. The peaks at 14.3°, 33.8°, and 57.5° are attributed to the (002), (100), and (110) plane of MoS2 crystal. From Raman spectroscopy shows the characteristic peaks of graphene (D, G and 2D) and MoS_2 (E^{1}_{2g} band at 377 cm⁻¹ and A_{1g} band at 403 cm⁻¹) are retained in the Raman spectra of G/MoS₂ which can confirm the fact that the hybrid of G/MoS₂ is composed of MoS₂ and graphene. Next, the XPS analysis was carried out to deduce the exact elemental composition of the G/MoS₂. The full scan of the G/MoS₂ gives the characteristic peaks for Mo 3d, S 2p, C 1s and O 1s with their corresponding binding energies. The morphologies and microstructures of the MoS₂, graphene and G/MoS₂ are systematically characterized by FESEM observation. The high resolution of FESEM image further reveals that the MoS₂ structures are constructed with layers of nanosheets. Meanwhile, FESEM image of graphene sheets illustrating the uniformly distributed of graphene into the Ni foam. Also, the inclusion of MoS₂ nanosheets resulted in a rough surface, logically due to co-stacking of MoS₂ nanosheets over the graphene nanosheets. Further, the morphology of the G/MoS₂ was examined by TEM and reveals the crystal lattice structure of MoS₂ and graphene in G/MoS₂. The interlayer spacing of MoS₂ in the hybrid were estimated to be ~0.63 nm, which can be indexed to their (002) lattice planes of hexagonal phase of MoS₂. Regardless of the difference in electrode being used, cyclic voltammetry (CV) analysis from the supercapacitor depicted a relatively good specific gravimetric capacitance (C_{sp}) and rate capability performance. A nearly rectangular-shaped CV curve was observed even at high scan rate. Besides, from the charge-discharge measurement, the symmetrical triangular curves reveal that there is no IR drops or voltage drops because of low internal resistance in the electrodes. Also, the electrode shows excellent discharge behavior and good capacitance retention of up to 10,000 cycles. Thus, this 2D heterostructures may provide excellent rate capabilities, high capacitance, and long lifecycle energy device. This is very promising for the development of high energy and high power density of device for multi-scale applications or industries.

ABSTRAK

Superkapasitor adalah peranti tenaga yang sangat menarik kerana prestasi penyimpanan cas elektrik dan keupayaan kitaran hayat yang penting. Pembinaan sel superkapasitor terutamanya fabrikasi elektrod adalah penting untuk memastikan prestasi aplikasi yang hebat. Tujuan penyelidikan ini adalah untuk menyediakan elektrod molibdenum disulfida (MoS₂), grafin dan hybrid G/MoS₂ dan penggunaannya sebagai superkapasitor simetri dan asimetrik. Elektrod telah disediakan dengan menggunakan teknik campuran separa cair yang mudah dan senang. Dengan ini, XRD digunakan untuk menganalisis fasa kristal dan struktur grafin yang disediakan, MoS₂, dan hibrid G/MoS₂. Puncak pada 14.3°, 33.8° dan 57.5° dikaitkan dengan kristal MoS₂ (002), (100), dan (110). Dari spektroskopi Raman menunjukkan puncak ciri-ciri grafin (D, G dan 2D) dan MoS₂ (band E^{I}_{2g} pada 377 cm⁻¹ dan Alg pada 403 cm⁻¹) dikekalkan dalam spektrum Raman hybrid G/MoS₂ yang dapat mengesahkan fakta bahawa hibrid G/MoS₂ terdiri daripada MoS₂ dan grafin. Seterusnya, analisis XPS dijalankan untuk menyimpulkan komposisi elemen sebenar hibrid G/MoS₂. Siaran penuh hibrid G/MoS₂ memberikan puncak ciri untuk Mo 3d, S 2p, C 1s dan O 1s dengan tenaga mengikat yang sama. Morfologi dan mikrostruktur MoS₂, grafin dan hibrid G/MoS₂ secara sistematik dicirikan oleh pemerhatian FESEM. Resolusi tinggi imej FESEM dengan lebih lanjut mendedahkan bahawa struktur MoS₂ dibina dengan lapisan nano. Sementara itu, imej FESEM lembaran grafin yang menggambarkan gambarajah secara seragam ke dalam Ni. Juga, kemasukan lapisan nano MoS₂ menyebabkan permukaan yang kasar, secara logiknya disebabkan oleh penyambungan lapisan nano MoS₂ ke atas lapisan nano grafin. Tambahan lagi, morfologi hibrid G/MoS2 diperiksa oleh TEM dan mendedahkan struktur kristal MoS2 dan grafin dalam hibrid G/MoS2. Jarak lapisan dalam MoS2 di dalam hibrid adalah dianggarkan ~ 0.63 nm, yang boleh diindeks ke permukaan (002) fasa heksagon MoS₂. Tanpa mengira perbezaan elektrod yang digunakan, analisa voltammetri siklik (CV) dari superkapasitor menggambarkan kapasitan gravimetrik spesifik (C_{sp}) dan prestasi kadar keupayaan yang agak baik. CV berbentuk persegi panjang hampir diperhatikan walaupun pada kadar imbasan tinggi. Di samping itu, dari pengukuran caj pelepasan, lengkung segitiga simetri mendedahkan bahawa tiada kejatuhan IR atau penurunan voltan kerana rintangan dalaman yang rendah dalam elektrod. Juga, elektrod menunjukkan tingkah laku pelepasan yang sangat baik dan pengekalan kapasiti yang baik sehingga 10,000 pusingan. Jadi, heterostruktur 2D ini boleh memberikan keupayaan kadar yang sangat baik, kapasitan yang tinggi, dan peranti tenaga kitaran hayat yang panjang. Ini sangat menjanjikan untuk pembangunan tenaga tinggi dan ketumpatan kuasa tinggi untuk aplikasi pelbagai skala atau industri.

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LIST OF ABBREVIATIONS

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| AC | - | Activated Carbon | |
|---------------------------------|---|---|--|
| ASC | - | Asymmetric Supercapacitor | |
| CNT | - | Carbon Nanotube | |
| СР | - | Conducting Polymer | |
| $C_{ m sp}$ | - | Specific Gravimetric Capacitance | |
| CV | - | Cyclic Voltammetry | |
| CVD | - | Chemical Vapor Deposition | |
| ED | - | Energy Density | |
| EDLC | - | Electrochemical Double Layer Capacitor | |
| EIS | - | Electrochemical Impedance Spectroscopy | |
| FESEM | - | Field Emission Scanning Electron Microscopy | |
| G | - | Graphene | |
| GCD | - | Galvanostatic Charge-Discharge | |
| H_2SO_4 | - | Sulfuric Acid | |
| KOH | - | Potassium Hydroxide | |
| LIB | - | Lithium Ion Battery | |
| MoS_2 | - | Molybdenum Disulfide | |
| Na ₂ SO ₄ | - | Sodium Sulfate | |
| NMP | - | N-Methyl-2-pyrrolidone | |
| PTFE | - | Polytetrafluoroethylene | |
| TEM | - | Transmission Electron Microscopy | |
| TMD | - | Transition Metal Dichalcogenide | |
| XRD | - | X-Ray Powder Diffraction | |
| XPS | - | X-Ray Photoelectron Spectroscopy | |
| 2D | - | Two Dimensional | |
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LIST OF SYMBOLS

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| F g ⁻¹ | - | Farad per gram |
|---------------------|---|-------------------------|
| g | - | Gram |
| k | - | Kilo |
| m | - | Meter |
| e | - | Electron |
| mA | - | MiliAmpere |
| mV s ⁻¹ | - | MiliVolts per second |
| μm | - | Micrometer |
| nm | - | Nanometer |
| Pa | - | Pascal |
| S | - | Second |
| W kg ⁻¹ | - | Watts per kilogram |
| Wh kg ⁻¹ | - | Watts hour per kilogram |
| °C | - | Degree celcius |
| V | - | Voltage |
| W | - | Watt |
| CV | | Cyclic voltammetry |
| Ψ0 | - | Electrode potential |
| Ψ | - | Potential |
| | | |

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(i) Peer reviewed journals

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(ii) Conference presentation

- Electrochemical performance of molybdenum disulfide Supercapacitor Electrode in potassium hydroxide and sodium sulfate electrolytes, 5th International Conference and Exhibition on Sustainable Energy and Advanced Materials Holiday Inn Melaka, 16–19 October 2017 (Oral presentation).
- 2. Effects of PTFE binder ratio on the performance of graphene supercapacitor, ISoRIS'17, Ramada Hotel, Melaka, 18 & 19 July 2017 (Oral presentation).

LIST OF AWARDS

- 29th International Invention, Innovation & Technology Exhibition 2018
 Place: KLCC Convention Centre Kuala Lumpur
 Date: 10–12 May 2018
 ITEX 2018 Silver medal
- UTeM Research and Innovation Expo (UTeMEX) Date: 17 November 2017 Place: FTK Universiti Teknikal Malaysia Melaka Jury Special award & Gold award
- Seoul International Invention Fair (SIIF 2016)
 Date: 1–2 December 2016, Korea
 Place: Seoul, Korea
 Silver Medal
- 4. Malaysia Technology Expo 2016 (MTE 2016)
 Date: 18–20 February 2016
 Place: PWTC Kuala Lumpur
 Special Award & Gold Award

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CHAPTER 1

INTRODUCTION

1.1 Background

Currently new electrode materials are under intense study to be used in supercapacitor applications, including transition metal oxides, metal hydroxides, and conducting polymers. Compared with other electrode materials such as carbon based materials, these electrode materials give high capacitance. Despite having high capacitance, these electrode materials possess low electrical conductivity, leading to inferior cycling stability as well as lower energy and power densities.

Graphene, composed of carbon atoms arranged in a honeycomb lattice, has been shown to possess unique properties. Additionally, the transition metal chalcogenide molybdenum disulfide (MoS₂) (Figure 1.1) shows prospects for various applications, including supercapacitors, because of the unique atomic structure analogues with graphene (Cao et al., 2013; Hu et al., 2013; Shi et al., 2013; He and Que, 2016).



Figure 1.1: Molybdenum disulfide atomic structure (He and Que, 2016)

Research into green and renewable energy materials is expanding, as well as the market for low cost and light-weight electrochemical energy storage systems. Among energy

storage devices, supercapacitors are one of the most popular energy storage devices contrasted with batteries because of their long life stabilities, fast charge-discharge processes, and high power density (Lu et al., 2013a).

Fabrication of superior electrode materials with well design structures is a crucial factor to increase high electrochemical performance of the energy storage devices such as high energy density an

d high power density (Cakici et al., 2017; Chen et al., 2017). Layered MoS_2 is analogous to the two-dimensional (2D) structure of graphene. The intercalation of electrolyte ions is permitted from the usage of MoS_2 as an electrode material in supercapacitors.

Thus, the key factor that determines the storage mechanism of supercapacitor is the diameter of electrolyte ion as well as matching of the interlayer spacing (Zhou et al., 2017). The construction of 2D hybrid heterostructures of graphene and MoS_2 (G/MoS₂) is believed to allow strongly coupled nanohybrid materials with optimized functionalities for supercapacitor application.

In addition, chemical vapor deposition (CVD), reduction-induced in situ selfassembly, chemical assembly, and the hydrothermal method (Figure 1.2) are simple methods which provide a unique approach for the synthesis of G/MoS_2 for energy storage applications (Yang et al., 2017). These novel materials and the fabrication of symmetric and asymmetric supercapacitors are important in meeting energy and power demands.

Asymmetric supercapacitors (ASCs) are composed of two dissimilar electrodes such as carbon and pseudocapacitive materials based electrodes (Wu et al., 2014; Sun et al., 2015). Commonly, the negative electrode in ASC is composed of carbon based materials including graphene, carbon nanotube, and activated carbon. Meanwhile, metal oxides and conducting polymers are the example of the positive electrode in ASC.